On the Diffuse Scattering of X-Rays by Single Crystals of Tetranitromethane¹⁾

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Introduction

Oda, Iida, Watanabé and Nitta^{3,4,5)} reported the crystal structure of cubic tetranitromethane using Laue and oscillation photographic data $(a=7.08\,\text{\AA}\ Z=2,\ T^3\text{d}-I\,\overline{4}3\,m)$. The observed high symmetry of the crystal, $T^3\text{d}-I\,\overline{4}3\,m$, was explained by a statistical arrangement of the atomic groups or molecules within the crystal. In other words, the central carbon atoms of the two molecules in the unit cell lie at the positions 000 and $\frac{1}{2}\,\frac{1}{2}\,\frac{1}{2}$, and the eight nitrogen atoms on the

four body-diagonals with the C—N bond axes are parallel to each of the other molecules, the orientation of the pairs of oxygen atoms in the nitro groups being, however, in certain disorder around the respective C—N bond axis. The simplest model will be the one in which all the pairs of oxygen atoms rotate freely around the C—N bond axes. A calculation^{5)*} was carried out based on this model and the result explains well the observed intensities of Bragg reflections and the circular diffuse halo, the maximum occurring at 32° for Cu K_{α} . In addition to

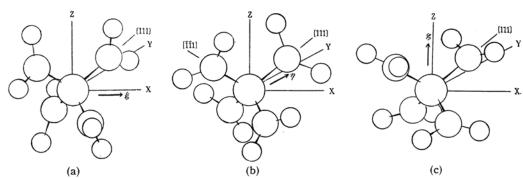


Fig. 1. The three different orientations of molecules in the crystal

$$N(K^{o}-K') = Nf^{j}{}_{o}\left\{8 + \sum_{n}^{1,2,3,4} 2J_{0}(2c_{n}) - \sum_{n}^{1,2,3,4} 4J_{0}^{2}\left(c_{n}\right)\right\}, \qquad (1)$$
 in which N is the number of molecules, f_{o} the atomic structure factor for oxygen, $J_{0}(c_{n})$ the Bessel function of the zeroth order with the argument c_{n} , c_{n} being
$$c_{n} = 4\pi(\rho/\lambda) \text{ sin } (\pi/2) \text{ sin } (\alpha_{n}),$$

where χ is the scattering angle, $a_{\rm n}$ the angle between the reflecting plane and the plane of the two rotating oxygen atoms in the n-th nitro group in the molecule, ρ the radius of the oxygen orbit. It may be added here that this equation is essentially the same as the one (4.237) in Zachariasen's "Theory of X-Ray Diffraction in Crystals", 1945 p. 224.

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^{*)} The calculation has given the x-ray diffuse scattering expressed by

these spot-like diffuse scattering was observed. In this paper we will show a more detailed analysis of the structure taking into consideration the mutual correlation between molecules, which enables us to explain such spot like diffuse scattering.

Theoretical

We shall assume, to simplify matters, a flatshaped molecule having the symmetry $D_{2d}-42 m$ (Fig. 1 (a), (b), (c)) which is not very different from the model given by Stosick⁶⁾ from his electron diffraction study on gaseous tetranitromethane. We shall take the X-axis in Fig. 1 (a), which is the four-fold alternating axis of the symmetry $D_{2d}-\overline{4}2 m$, as the molecular axis. It can be shown that by the simultaneous rotation of the four nitro groups, the molecular axis will be brought into Y-(Fig. 1 (b)) or Z-axis (Fig. 1 (c)), the positions of the nitrogen atoms remaining unchanged.

Let us denote by $P_l(x_0x_l)$ the probability that the orientational character of the molecule at the *l*-th lattice point will be of the kind x_l , when the one at the origin is of the kind x_0 . We shall normalize $P_l(x_0x_l)$ as

$$\sum_{x_0} P_l(x_0 x_l) = \sum_{x_l} P_l(x_0 x_l) = 1$$
 .

Since, in the present case, x is assumed to possess three possibilities corresponding to molecular orientation in the directions of X, Y and Z axes, we shall designate them further as ξ , η and ζ . As the averaged crystal structure is a bodycentered one, there are eight nearest lattice points about the l-th lattice point, which we shall designate as 1, 2, 3,..., μ ,...,8. The probability $W(x_1x_2\cdots x_{\mu}\cdots x_8;x_l)$ that the l-th is of the kind x_l if the ones surrounding it are of the kinds $x_1, x_2, \dots, x_{\mu}, \dots, x_8$ is now introduced. Then

$$P_{l}(x_{0}x_{l}) = \sum_{x_{1}}^{\xi, \eta, \zeta} \cdots \sum_{x_{8}}^{\xi, \eta, \zeta} \prod_{\mu=1}^{8} P_{\mu}(x_{0}x_{\mu})W(x_{1}x_{2}\cdots x_{\mu}\cdots x_{8}; x_{l}) \cdots (2)$$

When correlation is not large, $P_l(x_0x_l)$ will take a finite value, 1/3, for large value of l. Introducing,

$$P_l(x_0x_l) = (1/3) + R_l(x_0x_l),$$

and using the normalization condition for $W(x_1...$ $x_{\mu}\cdots x_8$; x_l),

$$\sum_{x_1} \cdots \sum_{x_8} \left(\frac{1}{3}\right)^8 W(x_1 \cdots x_{\mu} \cdots x_8; x_l) = \frac{1}{3},$$

it follows that, dropping the suffix zero of x_0 ,

$$R_l(x \ x') = \sum_{\mu=1}^{q} \sum_{x''}^{\xi, \eta, \zeta} R_{\mu}(x \ x'') A_{\mu}(x'' \ x') + N_l(x \ x'), \cdots (3)$$

$$A_{\mu}(x'' \ x') = \sum_{x''_1} \cdots \sum_{x''_{\mu-1}} \sum_{x''_{\mu+1}} \cdots \sum_{x''_8} \left(\frac{1}{3}\right)^{8-1} W$$

$$(x''_1 x''_2 \cdots x''_{\mu-1} x'' x''_{\mu+1} \cdots x''_8; x'),$$
where $\sum_{x''_1} \cdots \sum_{x''_8} \sum_{x''_8} means$ that the summation is to be

carried out over three orientations for each of

the seven molecules surrounding the l-th lattice point, the µ-th neighbouring molecule being fixed to the orientation x''. In other words, $A_{\mu}(x''x')$ means the probability that the molecule at the l-th lattice point is of the kind x' when the one at the μ -th neighbouring point is of the kind x''. $N_l(x|x')$ is the non-linear term of $R_l(x|x')$.

Expressing the potential energy between two neighbouring molecules by $V_{\mu}(x''x')$, it is easily shown by symmetry consideration that

$$\begin{split} &V(\xi,\xi)\!=\!V(\eta,\eta)\!=\!V(\zeta,\zeta)\\ &V(\xi,\eta)\!=\!V(\xi,\zeta)\!=\!V(\eta,\xi)\!=\!V(\eta,\zeta)\!=\!V(\zeta,\xi)\!=\!V(\zeta,\eta). \end{split}$$

Since $A_{\mu}(x''x')$ is expected to be proportional to the Boltzman factor $\exp\{-V_{\mu}(x''x')/kT\}$, we may assume that

$$\begin{split} &A(\xi,\xi) = A(\eta,\eta) = A(\zeta,\zeta) = (1+2\alpha)/3, \\ &A(\xi,\eta) = A(\xi,\zeta) = A(\eta,\xi) = A(\eta,\zeta) = A(\zeta,\xi) \\ &= A(\zeta,\eta) = (1-\alpha)/3, \end{split}$$

where α is a function depending upon potential and temperature.

Let $\mathbf{b} = (\sigma - \sigma_0)/\lambda$ where λ is the wave length of the x-rays, σ_0 and σ the unit vectors expressing the directions of the incident and scattered rays. The averaged intensity of scattered x-rays will be given by

$$\langle I(\mathbf{b}) \rangle_{\text{Av.}} = \sum_{l} \sum_{l'} \sum_{x_l} \sum_{x_{l'}} (1/3) P_{l'-l}(x_l x_{l'}) f(x_l) f^*(x_{l'})$$

$$= \sum_{l} \{2\pi i (\mathbf{b}, \mathbf{r}_{l'})\}$$

$$\dots \dots (4)$$

where $f(x_l)$ represents the "molecular" structure factor for the molecule at the l-th lattice point and \mathbf{r}_l its position vector, $\mathbf{r}_{l} t'$ being $\mathbf{r}_l - \mathbf{r}_{l'}$. This equation can be put into the following form7),

$$< I(\mathbf{b}) >_{Av} = I_1(\mathbf{b}) + I_2(\mathbf{b}), \quad \cdots (5)$$

in which

$$I_1(\mathbf{b}) = \operatorname{Spur}(\mathbf{F} S_0), I_2(\mathbf{b}) = \operatorname{Spur}(\mathbf{F} S_1), \dots (6)$$

$$S_0 = \left(\frac{1}{3}\right)^2 I \sum_{l \ l'} \exp\left\{2\pi i(\mathbf{b}, \mathbf{r}_{ll'})\right\} \qquad \cdots (7)$$

$$S_1 = \frac{1}{3} N \sum_{l} R_l \exp \left\{-2\pi i (\mathbf{b}, \mathbf{r}_l)\right\}, \qquad \cdots$$
 (8)

I being a matrix whose elements are all one, R_l a matrix whose elements are $R_l(x x')$ and N the number of molecules in the crystal. F is a matrix whose elements are products of two "molecular" structure factors for the various orientations. In the present case

$$\boldsymbol{F} = \begin{pmatrix} f \varepsilon^* f \varepsilon & f \varepsilon^* f \eta & f \varepsilon^* f c \\ f \eta^* f \varepsilon & f \eta^* f \eta & f \eta^* f c \\ f c^* f \varepsilon & f c^* f \eta & f c^* f c \end{pmatrix}$$

In (5) I_1 stands for the Laue-Bragg scattering and I_2 for the diffuse scattering. Since R_l tends to zero for large value of l, it is possible8) to calculate (8) and (6) by an approximate estimation of R_l for small values of l.

From (3) one finds, by introducing matrices A_{μ} and N_{l} whose elements are $A_{\mu}(x'' x')$ and $N_l(x\,x')$ respectively, the following expression,

$$R_l = \sum_{\mu=1}^{8} R_{\mu} A_{\mu} + N_l$$

or, noting that $R_{\mu}I=0$,

⁶⁾ A. J. Stosick, J. Am. Chem. Soc., 61, 1127 (1939).

⁷⁾ T. Matsubara, X-Rays, 6, 15 (1950).
8) T. Oda, X-Rays, 5, 95 (1949).

$$R_l = \sum_{\mu} R_{\mu} A'_{\mu} + N_l, \qquad \cdots (9)$$

where

$$A'_{\mu} = A_{\mu} - \frac{1}{3} I$$

In the present case, A'_{μ} is independent of μ and equal to

$$A' = \frac{\alpha}{3} \quad \begin{array}{cccc} 2 & -1 & -1 \\ -1 & 2 & -1 \\ -1 & -1 & 2 \end{array}$$

The summation over l of (9) multiplied by $\exp \{-2\pi i(\mathbf{b}, \mathbf{r}_l)\}$ gives

$$S_1 = NM(E-B)^{-1}, \qquad \dots (8')$$

where E is the unit matrix, and

$$B = \sum_{\mu=1}^{8} \varepsilon_{\mu} A'_{\mu}$$
, $\varepsilon_{\mu} = \exp\{-2\pi i (\mathbf{b}, \mathbf{r}_{\mu})\}$,

 \mathbf{r}_{μ} being a position vector expressing one of the eight lattice points measured from their central point, and

$$M = \frac{1}{3} \sum_{l} N_{l} \exp\{-2\pi i (\mathbf{b}, \mathbf{r}_{l})\}.$$
(10)

In this case B is given by

$$B=8(\cos\pi H\cos\pi K\cos\pi L)A'$$
,

where $H=(\mathbf{b}, \mathbf{a}_1)$, $K=(\mathbf{b}, \mathbf{a}_2)$, $L=(\mathbf{b}, \mathbf{a}_3)$ corresponds to a point in the reciprocal space, \mathbf{a}_1 , \mathbf{a}_2 , \mathbf{a}_3 being the cubic vector set.

By transforming \boldsymbol{B} into the diagonal matrix, $I_2(\mathbf{b})$ is expressed as follows:

$$I_3(\mathbf{b}) = N \sum_{i=1}^{3} \frac{m_{ii}}{1 - \lambda_i}$$

where $\lambda_1=0$, $\lambda_2=\lambda_3=8\alpha\cos\pi H\cos\pi K\cos\pi L$, and m_{ii} corresponds to diagonal elements of the matrix $UFMU^{-1}$ or UFU^{-1} UMU^{-1} , U being unitary and

$$U = \begin{pmatrix} 1/\sqrt{3} & 1/\sqrt{3} & 1/\sqrt{3} \\ 1/\sqrt{2} & 0 & -1/\sqrt{2} \\ 1/\sqrt{6} & -2/\sqrt{6} & 1/\sqrt{6} \end{pmatrix}$$

which has been used in the diagonalization of ${\it B.}$ ${\it UFU^{-1}}$ is calculated as

If we neglect all higher terms except the one for $l\!=\!0$ in (10), ${\it M}$ and ${\it UMU}^{-1}$ are given as follows:

$$M = (1/3)N_0$$
 and $N_0 = R_0 - \sum_{\mu=1}^{9} R_{\mu}A'$.

 R_0 can be written as

$$R_0 = E - (1/3)I = \frac{1}{3} \begin{pmatrix} 2 & -1 & -1 \\ -1 & 2 & -1 \\ -1 & -1 & 2 \end{pmatrix}$$

Hence, if we take $R_{\mu}=R_0A'$, UMU^{-1} becomes

$$UMU^{-1} = \begin{array}{ccc} 0 & 0 & 0 \\ 0 & m'_{22} & 0 \\ 0 & 0 & m'_{33} \end{array},$$

where $m'_{22} = m'_{33} = (1 - 8\alpha^2)/3$. From these calculations it follows that

$$I_2(\mathbf{b}) = N\left(\frac{1}{3}\right)^2 (1 - 8\alpha^2)$$

$$\frac{f \varepsilon - f_{\eta}^{-2} + |f \varepsilon - f_{\zeta}|^{2} + |f_{\eta} - f_{\zeta}|^{2}}{1 - 8\alpha \cos \pi H \cos \pi K \cos \pi L}, \quad \cdots (11)$$

where

$$\begin{aligned} & f \varepsilon - f_{\eta} = f_{0} \{ (C \varepsilon - C_{\eta}) + i (S \varepsilon - S_{\eta}) \}, \\ & f \varepsilon - f_{\zeta} = f_{0} \{ (C \varepsilon - C_{\zeta}) + i (S \varepsilon - S_{\zeta}) \}, \\ & f_{\eta} - f_{\zeta} = f_{0} \{ (C_{\eta} - C_{\zeta}) + i (S_{\eta} - S_{\zeta}) \}, \end{aligned}$$

and

 $C_{\zeta} = 4 \cos 2\pi Lz (\cos 2\pi Hx \cos 2\pi Ky + \cos 2\pi Kx \cos 2\pi Hy)$

 $S_{\zeta} = -4 \sin 2\pi Lz (\sin 2\pi Hx \sin 2\pi Ky + \sin 2\pi Kx \sin 2\pi Hy),$

 f_0 being the atomic structure factor for oxygen atoms and x, y, z its parameters.

When $\alpha = 0$, $I_2(\mathbf{b})$ becomes

$$N(1/3)^2 \{ f_{\xi} - f_{\eta} |^2 + |f_{\xi} - f_{\zeta}|^2 + |f_{\eta} - f_{\zeta}|^2 \}, \dots (12)$$

$$N\{(1/3)(|f_{\xi}|^2+|f_{\eta}|^2+|f_{\zeta}|^2)-|(f_{\xi}+f_{\eta}+f_{\zeta})/3|^2\}.$$

This last equation is equivalent to

$$N\{\langle f(x) | ^2 \rangle_{Av.} - |\langle f(x) \rangle_{Av.} | ^2 \},$$

which was derived in the previous paper⁵⁾ dealing with free axial rotation of nitrogroups. These equations account well for the observed halo-type diffuse scattering.

Now we will discuss the case where $\alpha \neq 0$. Since there is no long-range order with respect to molecular orientation, the correlation between neighbouring molecules would be small, in other words, the value of $|\alpha|$ will not be large. So we may assume 1>8 α . The sign of α depends upon the interaction energy V(xx') between two neighbouring molecules. For example if $V(\xi, \xi) > V(\xi, \eta)$, then $\alpha < 0$, and on the other hand if $V(\xi, \xi) < \langle \xi, \eta \rangle$, then $\alpha > 0$.

Since the function

$$|f_{\xi}-f_{\eta}|^2+|f_{\xi}-f_{\zeta}|^2+|f_{\eta}-f_{\zeta}|^2$$
(12')

changes slowly with **b**, one finds that the sharp maxima of I_2 (**b**) occur in the directions for which $1-8\alpha\cos\pi H\cos\pi K\cos\pi L$ possesses minimum value. When $\alpha<0$, such

directions are given by H, K, L with integers and H+K+L=2n+1. When H, K, L are integers, we have H=h, K=k and L=l. However diffuse spots associated with $\{110\}$, and $\{112\}$ (Table I) appeared. Hence, the as-

TABLE I
OBSERVED INTENSITIES OF MAXIMA OF THE
DIFFUSE SPOTS WITH THE DIRECTIONS $H=h,\ K=k$ AND L=l AND THEIR

	CALCULATED VALUES	
HKL	Obs.	Calc*
110	st.	29.6
200	w.	0.9
112	st.	19.2
220	w.	12.0

* In the calculation the polarization and the usual temperature fators have been also taken into account, B in $\exp\{-B\sin^2(X/2)\}$ being given as 3.

sumption that α is negative leads to disagreement with the observation, provided that the observed diffuse spots are to be explained only by the correlation of the orientations of molecules, disregarding the influence of thermal motions. On the other hand if we take $\alpha>0$, then the maxima of I_2 (b) occur for the directions b(corresponding to h+k+l=2n), in agreement with the observation.

We may assume that the potantial energies $V(x\,x')$ depend primarily upon the interaction between oxygen atoms of neighbouring molecules. The calculation of the distances between these oxygen atoms was carried out, taking the bond distances of C—N, N—O as

1.47, 1.21 \mathring{A} and the bond angle $\overset{N}{\underset{O}{\nearrow}}$ as

127°. The results are shown in Table II. In this case, we could expect that the potential energies for the two kinds of mutual orientations would not differ very much, so that $|\alpha|$ would be of a small value. It is, however, difficult to decide conclusively from such a qualitative consideration which one of these

TABLE II
CALCULATED VALUES OF THE INTERMOLECULAR.
ATOMIC DISTANCES BETWEEN OXYGEN

	ATOMS	
I		II
2.73 Å		2.92 Å
2.73		3.12
3.69		3.44
3.69		3.67
3.96		3.69
3.96		3.96

I stands for the parallel orientation ξ, ξ and II for the perpendicular ξ, η .

two is more stable and has a higher probability.

In the theory stated above, it is seen that when $\alpha>0$, the maxima depend upon the values of (12'). In Table I are tabulated the calculated values of this equation, with the parameter values x=0.28, y=0.11, z=0.16 which are close to those x=0.2675, y=0.054, z=0.1607 derived from the bond-lengths and bond-angles as mentioned above. As is seen the calculation explained qualitatively the observed results. Further approximation will be attained by examination of the more appropriate model^{4,5)} of the molecule having the symmetry S_4 — $\overline{4}$ and by taking into consideration of the lattice vibrations⁹⁾.

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